PO1: Core competency

PO2: Disciplinaryknowledgeandskill

PO3: Skilled communicator

PO4:Critical thinker and problem solver

PO5: Sense of inquiry

PO6: Team player

PO7: Skilled project manager

PO8: Digitally literate

PO9: Ethical awareness/reasoning

CC1: Fundamentals of Chemistry - I

(Credit: Theory -03, Practical - 01)

Course learning outcome (COs)

CO1: Extra nuclear structure of atoms and Periodicity

CO2: Basics of Organic Chemistry Bonding and Physical Properties:

Valence Bond Theory, Electronic displacement, MO Theory, Physical properties, Stereochemistry

CO3: Thermodynamics, Chemical Kinetics

CO4:Calibration and use of apparatus, Acid-Base Titrations, Oxidation-Reduction
Titrimetry

COs-POs Mapping

CC1: Fundamentals of Chemistry - I

	P01	P02	P03	P04	P05	P06	P07	P08	P09	P010
CO1	V	√		√	√				√	V
CO2	V	1		√	√				√	V
CO3	V	√		√	√				√	V
CO4	V	V	√	√	$\sqrt{}$	√			√	V

P01: Core competency

PO2: Disciplinaryknowledgeandskill

PO3: Skilled communicator

PO4:Critical thinker and problem solver

PO5: Sense of inquiry

P06: Team player

PO7: Skilled project manager

PO8: Digitally literate

PO9: Ethical awareness/reasoning

SEC1: Quantitative Analysis and Basic Laboratory Practices (Credit: Theory -03, Tutorial - 01)

Course learning outcome (COs)

CO1: Introduction to Quantitative analysis and its interdisciplinary nature

CO2: Titrimetric analysis, Acid-basetitrimetry, Redox titrimetry, Precipitation titrimetry, Complexometric titrimetry, Gravimetric Analysis

CO3: Water analysis, Water treatment technologies, Basic laboratory practices

CO4: Tutorial

COs-POs Mapping

SEC1: Quantitative Analysis and Basic Laboratory Practices

	P01	PO2	P03	PO4	P05	P06	P07	P08	P09	PO10
CO1	√	√		V	√				√	1
CO2	√	√		V	√				$\sqrt{}$	1
CO3	1	1		V	√				√	1
CO4	V	V	√	V	√		V		√	V

PO1: Core competency

PO2: Disciplinaryknowledgeandskill

PO3: Skilled communicator

PO4:Critical thinker and problem solver

PO5: Sense of inquiry

PO6: Team player

PO7: Skilled project manager

PO8: Digitally literate

PO9: Ethical awareness/reasoning

CC2: Fundamentals of Chemistry - II

(Credit: Theory -03, Practical - 01)

Course learning outcome (COs)

CO1: Kinetic Theory and Gaseous state, Real gas and Virial equation

CO2: Chemical Bonding - Ionic bond, Covalent bond

CO3: Stereochemistry, General Treatment of Reaction Mechanism, Reactive intermediates, Reaction thermodynamics, Reaction kinetics, Substitution Reaction

CO4:Iodo-/ Iodimetric Titrations, Estimation of metal content in some selective samples

COs-POs Mapping

CC2: Fundamentals of Chemistry - II

	P01	P02	P03	P04	P05	P06	P07	P08	P09	P010
CO1	1	1		√	√				√	√
CO2	V	V		√	√				√	√
CO3	V	V		√	√				√	√
CO4	V	√	√	V	V	V			V	√

P01: Core competency

PO2: Disciplinaryknowledgeandskill

PO3: Skilled communicator

PO4:Critical thinker and problem solver

PO5: Sense of inquiry

P06: Team player

PO7: Skilled project manager

PO8: Digitally literate

PO9: Ethical awareness/reasoning

SEC2: AI for Everyone

(Credit: Theory -04)

Course learning outcome (COs)

CO1: Introduction to Artificial Intelligence, Subfields and Technologies

CO2: Applications of AI and Ethical and Social Implications of AI

CO3: Other Important Issues like AI and Innovation Emerging trends and future directions in

ΑI

COs-POs Mapping

SEC2: AI for Everyone

	P01	PO2	PO3	P04	P05	P06	P07	P08	P09	PO10
C01	√	√	V	√	$\sqrt{}$			V		V
CO2	√	√	V		V			V		√
CO3	V	V	V		1		1	√		√

PO1: Core competency

PO2: Disciplinaryknowledgeandskill

PO3: Skilled communicator

PO4:Critical thinker and problem solver

PO5: Sense of inquiry

PO6: Team player

PO7: Skilled project manager

PO8: Digitally literate

PO9: Ethical awareness/reasoning

PO10: Lifelong learner

Chemistry MAJOR

(Credits:Theory-03,Practicals-01)

Paper: CHEM-H-CC3-3-Th /CHEM-H-CC3-3-P

Physical Chemistry -I

Course learning outcome(COs):

After going through the course the student should be able to understand

CO.1.Thermodynamics -II

CO.2. Application of thermodynamics -I

CO.3. Electrochemistry

CO.4. Determination of rate constant of different reactions experimentally

COs-POs Mapping

Course Code: CHEM-MD-CC3-3-Th(P)/CHEM-MD-CC3-5-Th (P)

	P01	PO2	P03	P04	P05	P06	P07	P08	P09	PO10
CO1	√	√	√	√	√					V
CO2	√	√	√	√	√					√
CO3	V	V	V	√	√					√
CO4	V	√	V	V	V					1

P01: Core competency

PO2: Disciplinaryknowledgeandskill

PO3: Skilled communicator

PO4:Critical thinker and problem solver

PO5: Sense of inquiry

P06: Team player

PO7: Skilled project manager

PO8: Digitally literate

PO9: Ethical awareness/reasoning

Chemistry MAJOR

(Credits:Theory-03,Practicals-01)

Paper: CHEM-H-CC4-3-Th /CHEM-H-CC4-3-P

Organic Chemistry -I

Course learning outcome(COs):

After going through the course the student should be able to understand

- CO.1.Aromatic substitution
- CO.2.General treatment of reaction mechanism-II
- CO.3. Substitution and elimination reactions
- CO.4. Chemistry of alkenes and alkynes
- CO.5. Identification of pure single organic compound (solid compounds, liquid compounds)

COs-POs Mapping

Course Code: CHEM-MD-CC3-3-Th(P)/CHEM-MD-CC3-5-Th (P)

	P01	PO2	P03	P04	P05	P06	P07	P08	P09	PO10
CO1	√	V	V	V	V					√
CO2	√	√	$\sqrt{}$	$\sqrt{}$	$\sqrt{}$					√
CO3	√		$\sqrt{}$	$\sqrt{}$						$\sqrt{}$
CO4	√		$\sqrt{}$	$\sqrt{}$						$\sqrt{}$
CO5	1		$\sqrt{}$	\checkmark						$\sqrt{}$

P01: Core competency

PO2: Disciplinary knowledge and skill

PO3: Skilled communicator

PO4: Critical thinker and problem solver

PO5: Sense of inquiry

P06: Team player

PO7: Skilled project manager

PO8: Digitally literate

PO9: Ethical awareness/reasoning

Interdisciplinary Course (IDC)

IDC-1 or IDC-2: Quantitative Analysis and Basic Laboratory Practices (Credit: Theory -03, Tutorial - 01)

Course learning outcome (COs)

CO1: Introduction to Quantitative analysis and its interdisciplinary nature

CO2: Titrimetric analysis, Acid-base titrimetry, Redox titrimetry, Precipitation titrimetry, Complexometric titrimetry, Gravimetric Analysis

CO3: Water analysis, Water treatment technologies, Basic laboratory practices

CO4: Tutorial

COs-POs Mapping

IDC-1 or IDC-2: Quantitative Analysis and Basic Laboratory Practices

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10
CO1		V		V	V					√
CO2		V		V	1					V
CO3		V		V	1				V	V
CO4		V	V	V	V		V			V